

87 % , + Fe^{3+} , 3-63 %.

(2).

2.

	As							
	0	20	40	80	100	120	140	160
1	0	4	20	7	40	10	80	
+	2	0	5	20	8	40	11	80
3	3	0	6	20	9	40	12	80
Fe^{3+}								
, 10				2		4		8

80 / (3).

Fe^{3+}

1,7-3,3

80 / + 3

3.

	/		,%	
			-	-
1-3	0	0	-	-
4	7,29	1,84	0	0
5	5,7	1,03	22	45
6	4,59	0,48	37	74
7	17,6	4,08	0	0
8	16,14	1,91	8	53
9	11,38	0,73	35	82
10	41,3	8,7	0	0
11	32,8	4,19	21	52
12	23,7	2,64	43	70
05	1,01	0,38		

,)%

: 1) (, 40 /
, 2) 40 /
, 3)

4)

+CaCO₃.

Fe^{3+}

4

40 /

Fe^{3+} ,

0,

2. 61 %, - 50 %

75 %, - 73, - 46 %.

4.

(/ , s)	/		
	0	0	0
40,	19,6	2,6	0,4
40, +	7,7	1,3	0
40, + 3	10,5	0,7	0,1
05	2,43	0,35	0,05

As (V)
[3, 4].

0,8 + 3 Fe^{3+} 0,2-0,4
- 0,1-

(6,9-7,5).

Fe^{3+}
CaCO₃

1.

01.01.2000 .) - : , 2002. - 30 .

2.

- 2009.- 7.- . 794-805.

3.

// - 2009.- 8.- . 85-94.

4.

: . - : , 1989. - 439 .

5.

- 2009.- 2. - . 16-18.

6.

2008. - 528 .

RECLAMATION OF ARSENIC-POLLUTED SOILS

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Effect of different natural sorbents, their mixtures, and modifications on the transfer of arsenic from mobile forms to inactive ones in the soil was studied. Composite mixtures increasing the efficiency of arsenic sorption in the soil were developed. Fe^{3+} -treated diatomite was found to best bind arsenic; it was mixed with blue clay to prevent its input to crops.

Keywords: soil, sorbent, composite mixtures, pollution, arsenic, diatomite, iron, blue clay, sapropel.